

# Low Range Test Tube Format Nitrate Test Kit

(0 to 1.0 ppm Nitrate-N)

Accurate, economical, and safe nitrate analysis.

- Analyze 25 samples
- Nitrate Standards included with the kit

Nitrate Units	US EPA	CA & Europe	Molarity
Low Range Kit	0.05 - 1.0 ppm Nitrate-N	0.2 – 4.4 ppm Nitrate	3.6 - 71.4 µM Nitrate

Nitrate is reported in different units depending on your field of use and where you live.

### **OVERVIEW**

- ✓ **Store kit refrigerated** or below 60°F (15°C).
- ✓ See tag for expiration date.

This kit will provide reliable estimates of nitrate content when used as supplied.

For **quantitative data**, measure reagents using pipets and read assay results with a colorimeter at 540nm.

# ♦ EQUIPMENT AND REAGENTS

You will need to supply:

- distilled or deionized water
- clean, nitrate-free containers if you are collecting samples to analyze later

We use the abbreviation "d-I water" for distilled or deionized water.

Superior Enzymes 334 Hecla Street

Lake Linden, Michigan 49945 Tech: 906.296.1130 Sales: 906.296.1115 This Nitrate test kit is based on the enzyme Nitrate Reductase (NaR), catalyzing the reduction of Nitrate to Nitrite using the natural electron donor NADH. The Nitrite reacts with color reagents (dyes) under acidic conditions to produce a visible color. The concentration of Nitrate in the original sample is determined by measuring absorbance versus Nitrate concentration in Nitrate Standards. Nitrate can be determined in water samples and extracts of plant tissues, soils and foods. The test is designed to measure Nitrate in the range of 0.05 to 1.0 ppm Nitrate-N in up to 25 samples or standards. The Nitrate concentration can also be expressed as  $\mu$ M Nitrate, where range is 3.6 to 71.4  $\mu$ M Nitrate, or ppm Nitrate, where range is 0.2 to 4.4 ppm Nitrate. Nitrite can also be determined by omitting NaR and NADH from the test (see Determining Nitrite, page 4).

If testing seawater, salt water or brackish water, follow the blue instructions.

Chloride is a mild inhibitor of Nitrate Reductase. The color development when analyzing seawater is not as intense as other types of samples. Nitrate Standards prepared in salt water and longer reaction times solve the problem. The longer reaction time (40 minutes) will produce more color development, making your nitrate determination more accurate.

### Supplied in NECi Test Kit

- □ Assay Buffer (AB) in liquid form one 50 ml tube
- □ Color Reagent No. 1 in solid form one 15 ml amber bottle
- □ Color Reagent No. 2 in solid form one 15 ml amber bottle
- □ NADH in freeze-dried form one tube in amber bag
- □ Nitrate Reductase (NaR) in freeze-dried form one tube in foil pouch
- □ Enzyme Diluent one squeeze-bulb
- □ Nitrate Standard (100 ppm Nitrate-N) in liquid form one 1.5 ml tube
- ☐ Microcentrifuge tubes six tubes for preparing Nitrate Standards
- □ Salt Water one 20 ml tube, green cap, only if you are testing seawater

# Supplied by User

- □ 25 or 100 ml graduated cylinder.
- □ Variable pipetters (10 to 100  $\mu$ l and 100 to 1000 $\mu$ l).
  - **Test tube vortex-type mixer** or other means to mix contents of tubes.
- □ Colorimeter or Spectrophotometer capable of reading at 540 nm± 20 nm, with a glass or plastic cuvette (approx. volume 2 ml).
- □ (25) 13 x 100 mm test tubes (Clean and Nitrate-free).
- $\blacksquare$  **Timer** (0 to 40 minutes) a clock or stop watch is adequate.
- □ **Deionized or distilled water** (d-I water; must be "Nitrate-free" to avoid high background).
- □ 4 ml of concentrated HCl.
- ☐ Ice and Ice Bucket.

NEED HELP? Contact NECi
Toll Free: 1-888-NITRATE FAX: 1-906-296-8003
Email: tech@nitrate.com

Visit us on the web: www.nitrate.com

# ◆ REAGENT PREPARATION

Step 1 Assay Buffer – ready to use from kit. Warm to room temperature for nitrate tests. If desired, the assay buffer may be more quickly warmed in a 30°C water bath.

Step 2 Prepare 3 N HCl by adding 4 ml concentrated HCl to 12 ml d-I water. Mix.

Step 3 Add 15 ml 3 N HCl to Color Reagent No. 1 bottle. Mix by shaking well.

Step 4 Add 15 ml d-I water to Color Reagent No. 2 bottle. Mix by shaking well.

Step 5 Remove tube of **NADH** from amber bag, tap tube to settle contents, add 1.5 ml **d-I water** and replace cap. Mix by inversion several times. Keep on ice during use.

Step 6 Remove NaR vial from foil pouch and tap tube to settle contents before opening. Twist off the end of the Enzyme Diluent Squeeze Bulb and completely empty the contents into the NaR vial. Replace the cap and mix by inversion 3 times. Allow to stand at room temperature for at least 10 minutes, with mixing at 5 and 10 minutes. Then keep on ice during use.

# ➤ **Assay Buffer** - 28 mM KH<sub>2</sub>PO<sub>4</sub>, 0.025 mM EDTA; pH 7.5.

- ➤ Color Reagent No. 1 1% Sulfanilamide in 3N HCl.
- ➤ Color Reagent No. 2 0.02% N-Naphthylethylenediamine in d-I water.
- ➤ **NADH** approx. 2 mM NADH.
- ➤ Nitrate Reductase (NaR) 0.5 unit of NaR per tube; 1.0 unit for Saltwater kit.
- ➤ **Nitrate Standard** 1 vial of 100 ppm nitrate-N.
- > Salt Water may contain 0.05 ppm nitrate-N, giving your blank a slight pink color.

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# STANDARD PREPARATION

Transfer 1 ml of 100 ppm Nitrate-N Standard into a test tube containing 9 ml d-I water to make a 10 ppm Nitrate-N Standard. In another test tube, dilute 1 ml of 10 ppm Nitrate-N Standard with 9.0 ml d-I water to make 1.0 ppm Nitrate-N Standard. Use the 6 microtubes (provided in kit) to prepare Nitrate Standards as shown in table below. Cap and mix the tubes by inversion before use.
If you are testing seawater, use the Salt Water provided instead of d-I water to prepare standards.

Vol 1 ppm Nitrate-N Standard	Volume d-I water	Resulting Standard (ppm Nitrate-N)	Resulting Standard (ppm Nitrate)	Resulting Standard (µM)
1000	0	1.0	4.4	71.2
750	250	0.75	3.3	53.4
500	500	0.5	2.2	35.6
250	750	0.25	1.1	17.8
100	900	0.1	0.6	7.1
50	950	0.05	0.2	3.6

NOTES ON THE

REAGENTS

## ♦ NITRATE ASSAY PROCEDURE

- The following procedure is written for single determinations.
- For greater accuracy, replicates can be run.

#### WASTE DISPOSAL

Follow all local guidelines and regulations. If there are no local guidelines, wash the waste down the sink with large amounts of running water.

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#### **♦ CALCULATIONS**

STEP 1 Pipette 500 μl d-I water into one test tube for use as reagent blank. If testing seawater, use the Salt Water provided instead of d-I water.

STEP 2 Pipette 500 μl of the samples and standards into the required number of test tubes.

STEP 3 Add 500 µl Assay Buffer to each tube. STEP 4 Add 50 µl NADH solution to each tube

Add **50 \mul NADH solution** to each tube. Mix thoroughly with a vortex-type mixer. If testing seawater, use only 25  $\mu$ l of NADH solution.

STEP 5 Add 40 µl NaR solution to each tube. Mix thoroughly with a vortex-type mixer. If testing seawater, use 40 µl of NaR solution.

STEP 6 Let tubes sit for ~20 minutes at room temperature. (NOTE: Exact timing is not critical but at least 20 minutes are required for complete reduction of nitrate.) If testing seawater, let tubes sit for at least 40 minutes.

STEP 7 Add 500 µl Color Reagent No. 1 to each tube. Mix thoroughly with a vortex-type mixer.

STEP 8 Add 500 µl Color Reagent No. 2 to each tube. Mix thoroughly with a vortex-type mixer.

STEP 9 Let tubes Stand at room temperature for ~10 minutes. To ensure homogeneous samples, briefly mix the tubes with a vortex-type mixer.

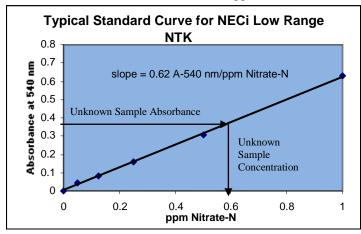
STEP 1 To correct for any background absorbance due to the reagents, subtract the mean absorbance of the reagent blank(s) from the mean absorbance of each nitrate standard and unknown sample:

Corrected mean sample A-540 nm = (mean A-540 nm for sample) – (mean A-540 nm for reagent blank)

Generate a standard curve for the Nitrate Standard (see example below).

Using linear graph paper or a computer plotting program such as Sigma Plot® or spreadsheet such as Excel®, plot the ppm Nitrate-N on the x-axis, and the A-540 nm for each nitrate standard on the y-axis. If plotting by hand, draw a straight line through the points for the Nitrate Standards. If plotting by computer, the slope of the line can be calculated for determining Nitrate-N ppm in the unknown samples.

Using the standard curve, determine the ppm Nitrate-N for the sample: (a) Find the corrected A-540 nm for the sample on the y-axis of the standard curve. (b) Follow over along a horizontal line to where the line intersects the standard curve. Trace down the x-axis and read the ppm of Nitrate-N on the x-axis.



# UNKNOWN SAMPLES WITH HIGH NITRATE

This NECi Nitrate Test Kit is capable of determining Nitrate levels of up to 1.0 ppm Nitrate-N (71  $\mu$ M Nitrate). If an unknown sample is found to have more than 1.0 ppm Nitrate-N, the sample may be diluted with d-I water 1:10 to allow an exact determination. For example, take 100  $\mu$ l of sample and add 900  $\mu$ l of deionized water to make a 1:10 dilution and then assay 500  $\mu$ l of the diluted sample. After finding the Nitrate content of the diluted sample, multiply the Nitrate concentration by 10 to find the Nitrate concentration in the original sample . NOTE: Keep the sample volume constant by diluting the sample rather than using a smaller volume of sample in the assay.

DETERMINING NITRATE IN MOLAR UNITS

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Use a conversion of 1ppm Nitrate- $N=71~\mu M$  Nitrate. One ppm Nitrate-N=1~mg of Nitrogen/liter. Since Nitrogen has a molecular weight of 14 g/mole, then the molar concentration is:

 $(0.001 \text{ g/1}) \div (14 \text{ g/mole}) = 0.000071 \text{ M Nitrogen} = 0.000071 \text{ M Nitrate} = 71 \mu\text{M Nitrate}$ 

DETERMINING NITRATE AS A QUANTITY

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(Using the 500  $\mu$ l Sample Size). In the tube where the 1.0 ppm nitrate-N standard is determined, there is approx. 36 nmol of Nitrate [(71 nmol Nitrate/ml) x (0.5 ml) = 35.5 nmol]. So the example standard curve would have a slope of 0.017 A-540 nm/nmol Nitrate (calculated from slope = 0.62 A-540 nm/35.5 nmol Nitrate).

NITRATE DETERMINATION IN PLANT LEAF EXTRACTS

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To determine how much Nitrate is in one gram of leaf tissue, grind the tissue in 10 ml d-I water and measure the total volume of extract after the solids are filtered off. Take 500 µl of extract for the Nitrate assay, conduct the Nitrate Test Kit assay and find the amount of Nitrate present in nmoles. Determine the total amount of Nitrate in the extract [= (total volume of extract) x (nmol Nitrate/500 µl of extract)]. Divide this total amount of Nitrate by the weight of plant tissue to find the amount of Nitrate per unit of tissue (grams in this example). The green or brown color of the leaf extract may significantly interfere with Nitrate determinations since the plant extract is diluted only 2-fold in the assay. Using activated charcoal may overcome the interference when quantitative analysis of leaf Nitrate content is done by boiling leaves for 20 min. After boiling, cool on ice and then filter the sample to recover aqueous extract. Finally, make the volume back up to 10 ml to compensate for water lost during boiling. Ask NECi for a detailed protocol for extracting either fresh plant leaves or dried leaf material for Nitrate Assays.

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# **Determining Nitrite**

Interested in determining Nitrite? Nitrite can be determined by omitting NaR and NADH from the samples. (That is, skipping steps 4, 5 and 6 on page 3). Prepare Nitrate standards as described in the normal Nitrate Assay Procedure with both NADH and NaR added and use the Nitrate Standard Curve for estimating Nitrite content.

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The Clean Water and Safe Drinking Water Acts (U.S. EPA 1974) set Maximum Contaminant Level (MCL) for potable water at 10 ppm Nitrate-N (10 mg Nitrate-N per liter). California and European standards are 45 ppm nitrate (45 mg nitrate per liter). If you find drinking water with 7 to 10 ppm Nitrate-N or more, advise users to seek a professional test of their water. Environmental water samples usually contain 1 to 2 ppm Nitrate-N or less.

in Water

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