



Nitrate by NADH Disappearance Lab Kit for Food Testing

(0 to 10 ppm Nitrate-N)

Accurate, economical, and safe nitrate analysis.

Analyze 100 samples

Nitrate Standards included with the kit

Nitrate Units	US EPA	CA & Europe	Molarity		
Standard Range Kit	0.5 - 10.0 ppm Nitrate-N	2.0 - 44 ppm Nitrate	36 - 714 µM Nitrate		
Nitrate is reported in different units depending on your field of use and where you live.					

OVERVIEW

- ✓ Store kit refrigerated or below 60°F (15°C).
- ✓ See box for expiration date.

This kit will provide reliable estimates of nitrate content when used as supplied.

For **quantitative data**, measure reagents using pipets and read assay results with a spectrophotometer at 340nm.

EQUIPMENT AND REAGENTS

You will need to supply:

- distilled or deionized water
- clean, nitrate-free containers if you are collecting samples to analyze later

We use the abbreviation "d-I water" for distilled or deionized water.



This Nitrate test kit is based on the enzyme Nitrate Reductase (NaR), catalyzing the reduction of Nitrate to Nitrite using the natural electron donor NADH. The concentration of Nitrate in the original sample is determined by measuring absorbance versus Nitrate concentration in Nitrate Standards. Nitrate can be determined in water samples and extracts of plant tissues, soils and foods. The test is designed to measure Nitrate in the range of 0.5 to 10 ppm Nitrate-N in up to 100 samples or standards. The Nitrate concentration can also be expressed as μ M Nitrate, where range is 36 to 714 μ M Nitrate, or ppm Nitrate, where range is 2.0 to 44 ppm Nitrate.

Prepare your food samples according to your standard procedures. Make sure your samples are ready to assay before starting the procedure on page 3.

Supplied in NECi Test Kit

- □ Assay Buffer (AB) in liquid form five 50 ml tubes
- □ NADH in freeze-dried form eight tubes in amber bag
- □ Nitrate Reductase (NaR) in freeze-dried form four tubes in foil pouch
- **Enzyme Diluent** four squeeze-bulbs
- □ Nitrate Standard (100 ppm Nitrate-N) in liquid form one 1.5 ml tube
- □ Microcentrifuge tubes six tubes for preparing Nitrate Standards
- □ **Cuvettes -** 7 for reactions, UV compatible

Supplied by User

- □ 10 ml graduated cylinder.
- **Variable pipetters** (10 to 100 μ l and 100 to 1000 μ l).
- □ **Spectrophotometer** capable of reading at 340 nm± 20 nm, with a glass or plastic cuvette (approx. volume 3.5 ml).
- **Timer** (0 to 20 minutes) a clock or stop watch is adequate.
- □ **Deionized or distilled water** (d-I water; must be "Nitrate-free" to avoid high background).
- □ Ice and Ice Bucket.
- □ Extra cuvettes may be needed. Make sure they are UV compatible (optical glass, plastic or quartz)

NEED HELP? Contact NECi Toll Free: 1-888-NITRATE FAX: 1-906-296-8003 Email: <u>tech@nitrate.com</u> Visit us on the web: <u>www.nitrate.com</u>

◆ REAGENT PREPARATION	Step 1			use from kit. Warm ter may be more quick		
	Step 2			H from amber bag, ta p. Mix by inversion	-	
	Step 3	Twist off contents in to stand at	the end of the E nto the NaR via	oil pouch and tap tub Enzyme Diluent Squ I. Replace the cap and sure for at least 10 mi are during use.	eeze Bulb and con nd mix by inversion	mpletely empty the on 3 times. Allow
NOTES ON THE REAGENTS		NADH – 1 Nitrate R	2.4 mg per tube. eductase (NaR) -	I ₂ PO ₄ , 0.025 mM EDT. – 1.0 unit of NaR per of 100 ppm nitrate-N.	-	
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◆ STANDARD PREPARATION	Transfer 1 ml of 100 ppm Nitrate-N Standard into a test tube containing 9 ml d-I water to make a 10 ppm Nitrate-N Standard. Use the 6 microtubes (provided in kit) to prepare Nitrate Standards as shown in table below. Cap and mix the tubes by inversion before use.					
	Vol	10 ppm		Resulting	Resulting	Resulting
	Nit	rate-N	Volume	Standard	Standard	Standard
	-	indard	d-I water 0	(ppm Nitrate-N) 10.0	(ppm Nitrate) 44	(µM) 712
		750	250	7.5	33	534
		500	500	5.0	22	356
		250	750	2.5	11	178
		100	900	1.0	4.4	71.2
		50	950	0.5	2.2	35.6

0

0

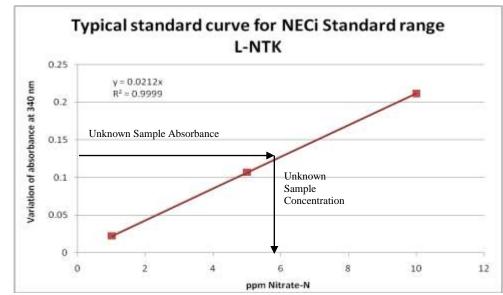
NITRATE ASSAY PROCEDURE cuvettes. The following STEP 3 Add 1.80 ml Assay Buffer to each cuvette. procedure is written for single determinations. For greater accuracy, replicates at 340nm (A1). can be run. WASTE DISPOSAL Follow all local guidelines and regulations. If there are no local guidelines, wash the waste down the sink with **STEP 8** Read A340 of each cuvette at exactly 30 minutes (A3). large amounts of running water. ***** CALCULATIONS STEP 2 in the unknown samples. STEP 3 L-NTK 0.25 y = 0.0212xVariation of absorbance at 340 nm $R^2 = 0.9999$ 0.2 Unknown Sample Absorbance 0.15 Unknown 0:1 Sample 0.05

STEP 1 Pipette **100 µl d-I water** into one cuvette for use as reagent blank.

- **STEP 2** Pipette **100 µl** of your prepared **samples** and **standards** into the required number of
- **STEP 4** Add **100 µl NADH solution** to each cuvette. Cap and mix thoroughly.
- **STEP 5** Zero the spectrophotometer with d-I water in a cuvette. Read absorbance of each cuvette
- **STEP 6** To start the reaction add **40 µl NaR** solution to each tube. Recap and mix thoroughly.
- **STEP 7** Read A340 of each cuvette at exactly 20 minutes (A2) after adding NaR.

STEP 1 Determine absorbance differences (A1-A2) and (A2-A3) for the blank and samples. Absorbance difference of the blank = $(A1-A2)_{blank} - 2 \times (A2-A3)_{blank}$ Absorbance difference of the sample = $(A1-A2)_{sample} - 2 x (A2-A3)_{sample}$ $\Delta A_{nitrate}$ = absorbance difference _{sample} – absorbance difference _{blank}

- Generate a standard curve for the Nitrate Standard (see example below). Using linear graph paper or a computer plotting program such as Sigma Plot® or spreadsheet such as Excel®, plot the ppm Nitrate-N on the x-axis, and the ΔA -340 nm for each nitrate standard on the y-axis. If plotting by hand, draw a straight line through the points for the Nitrate Standards. If plotting by computer, the slope of the line can be calculated for determining Nitrate-N ppm
- Using the standard curve, determine the ppm Nitrate-N for the sample: (a) Find the Δ A-340 nm for the sample on the y-axis of the standard curve. (b) Follow over along a horizontal line to where the line intersects the standard curve. Trace down the x-axis and read the ppm of Nitrate-N on the x-axis.



UNKNOWN SAMPLES WITH HIGH NITRATE	This NECi Nitrate Test Kit is capable of determining Nitrate levels of up to 10 ppm Nitrate-N (714 μ M Nitrate). If an unknown sample is found to have more than 10 ppm Nitrate-N, the sample may be diluted with d-I water 1:10 to allow an exact determination. For example, take 100 μ l of sample and add 900 μ l of deionized water to make a 1:10 dilution and then assay 100 μ l of the diluted sample. After finding the Nitrate content of the diluted sample, multiply the Nitrate concentration by 10 to find the Nitrate concentration in the original sample . NOTE: Keep the sample volume constant by diluting the sample rather than using a smaller volume of sample in the assay.
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DETERMINING NITRATE IN MOLAR UNITS	Use a conversion of 1ppm Nitrate-N = 71 μ M Nitrate. One ppm Nitrate-N = 1 mg of Nitrogen/liter. Since Nitrogen has a molecular weight of 14 g/mole, then the molar concentration is:
	$(0.001 \text{ g/1}) \div (14 \text{ g/mole}) = 0.000071 \text{ M Nitrogen} =$
	0.000071 M Nitrate = 71μ M Nitrate
****	******
DETERMINING NITRATE AS A QUANTITY	(Using the 100 μ l Sample Size). In the tube where the 10 ppm nitrate-N standard is determined, there is approx. 71 nmol of Nitrate [(714 nmol Nitrate/ml) x (0.10 ml) = 71.4 nmol]. So the example standard curve would have a slope of 0.0003 A-340 nm/nmol Nitrate (calculated from slope = 0.0212 A-340 nm/71.4 nmol Nitrate).
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FOODSAMPLE PREPARATION	Use your standard food sample preparation method prior to starting the nitrate assay procedure on page 3.

Thanks for using our products. Call Tech Support: 1.906.296.1130, or visit the NECi website: <u>www.nitrate.com</u> if you need more information. We're always interested in hearing about your experience with our kits.

NECi Superior Enzymes: Clean Water. Fertile Soil. Serious Science.

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